

POTENTIATION OF ANTIBACTERIAL EFFECTS OF DIFFERENT PYRIMIDINE ANALOGUES BY N-FORMYLBUIRET

A. ČIHÁK, J. HORÁK^a, J. BROUČEK and F. ŠORM

*Institute of Organic Chemistry and Biochemistry,
Czechoslovak Academy of Sciences, Prague 6*

Received May 13th, 1971

The combination of N-formylbiuret with some pyrimidine analogues of nucleic acid components results in a striking potentiation of their antibacterial effects. The analogues tested were 6-azapyrimidines, 5-azaorotic and 5-fluoroorotic acid, 5-fluorouracil, 5-fluorouridine and 5-fluoro-2'-deoxyuridine. The biological effects of other tested halogenopyrimidines, 5-azapyrimidines, antifolates, uracil and cytosine arabinosides, and of purine analogues were not affected by the simultaneous application with N-formylbiuret.

By studying the biological effects of N-formylbiuret, a degradation product of 5-azauracil^{1,2} it was established that this compound possesses antibacterial² and antimetabolic³ activity. Upon the simultaneous application of 6-azauridine and N-formylbiuret a pronounced potentiation of antibacterial², virostatic⁴ and partly cancerostatic⁵ effects of this compound was observed. A similar biological effect was found with other N-substituted and N,N'-disubstituted derivatives of biuret^{6,7}.

In the present communication we took up the possibility of potentiation of the biological effects of other, practically employed analogues of nucleic acid components, by using simultaneous application of N-formylbiuret. Together with this study, we attempted to localize the inhibitory action of this drug in cell metabolism.

EXPERIMENTAL

Reagents. 5- and 6-azapyrimidines were prepared at this Institute as described previously⁸. Similarly, N-formylbiuret⁷ was prepared here. Other pyrimidine and purine analogues were from Calbiochem, Los Angeles, and Hoffmann-La Roche, Nutley. L-Dihydroorotic acid, carbamyl phosphate and D,L-ureidosuccinic acid were from Calbiochem. L-Aspartic- [U-¹⁴C] acid (26 Ci/mol) was from the Institute for Research, Production and Uses of Radioisotopes, Prague.

Antibacterial activity. *Escherichia coli* B grown in a mineral synthetic medium² with glucose as a carbon source was used in the experiments. Cultivation was carried out under aseptic conditions at 37°C in test-tubes provided with Kapsenberg stoppers. Compounds dissolved directly before the experiment in the medium were added simultaneously, before inoculation of the culture with a drop of *E. coli* suspension.

* Present address: Laboratory for Cell Membrane Transport, Institute of Microbiology, Prague-Krč.

Cell-free extract of *E. coli* was prepared from a culture aerated for 8 h at 37°C in 5 l of synthetic medium. Washed cells were suspended under cooling and sonicated (MSE ultrasonic disintegrator, 80 s, 2°C) in 0.05M-Tris-HCl buffer, pH 7.4. Cell debris was centrifuged (10000 g, 20 min, 2°C) and the supernatant fraction was divided in small lots and maintained at -20°C. The protein content was determined according to Lowry and coworkers⁹.

Carbamyl aspartotranskinase was assayed at 30°C (total volume 1 ml) after adding of 0.2 ml of enzyme extract. The incubation (20 min) was carried out¹⁰ in $2 \cdot 10^{-2}$ M-Tris-HCl buffer of pH 8.7 in the presence of 10^{-2} M-Mg²⁺-ions, $2.5 \cdot 10^{-3}$ M carbamyl phosphate and 10^{-3} M L-aspartic-[U-¹⁴C] acid. The ureidosuccinic-[¹⁴C] acid formed was separated chromatographically on Whatman No 1 paper in 1-butanol-pyridine-acetic acid-water (15 : 10 : 3 : 12). Radioactivity was measured after elution on planchets in a gas-flow counter. The enzyme activity is expressed as μ mol of ureidosuccinic acid formed.

Dihydroorotase. The incubation was done¹¹ for 10 min at 37°C in a total volume of 5 ml in $2 \cdot 10^{-2}$ M-Tris-HCl of pH 8.0 after adding 0.3 ml of enzyme extract in the presence of $5 \cdot 10^{-3}$ M EDTA and $1.5 \cdot 10^{-3}$ M L-dihydroorotic acid. The enzyme activity was determined from the decrease of absorbance at 240 nm and is expressed as μ mol of ureidosuccinic acid formed. It is a disadvantage of the method that the ureidosuccinic acid formed undergoes simultaneous recycling to 5-carboxymethylhydantoin¹².

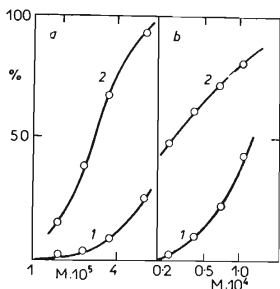


FIG. 1

Potentiation of Antibacterial Activity of 6-Azauracil (a) and 6-Azacytosine (b) by N-Formylbiuret

The compounds were added to the medium before inoculation either alone (1) or in combination with $2 \cdot 10^{-5}$ M N-formylbiuret (2) which does not affect bacterial growth at this concentration when present alone. Static cultivation of *E. coli* B for 16 h at 37°C. %, Inhibition of bacterial growth; the culture growth without inhibitor = 100%. M, Concentration of added analogue.

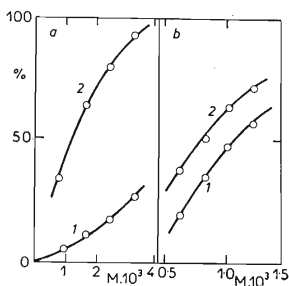


FIG. 2

Different Effects of N-Formylbiuret on the Antibacterial Activity of 6-Azauridine (a) and 5-Azauridine (b)

The compounds were added to the medium before inoculation either alone (1) or in combination with $2 \cdot 10^{-5}$ M N-formylbiuret (2). For details see Fig. 1.

RESULTS AND DISCUSSION

It was found that, as in case of 6-azauracil and 6-azauridine², simultaneous application of N-formylbiuret results in an enhanced antibacterial activity of 6-azacytosine and 6-azacytidine (Figs 1–3). The antibacterial effects of 6-azathymine and 6-azathymidine were not affected by N-formylbiuret. The mechanism of action of the thymine analogues differs¹³ from the inhibitory mechanism of potentiated 6-azapyrimidines which inhibit practically selectively the *de novo* synthesis of pyrimidines¹⁴. In contrast with 6-azapyrimidines, the antibacterial effect of the corresponding 5-azaanalogues¹⁵ is not affected by the presence of N-formylbiuret (Figs 1–3). It is known that these compounds, similarly to 6-azapyrimidines, interfere with the *de novo* synthesis of pyrimidines¹⁶. However, due to spontaneous and enzymic transformations and incorporation into different types of nucleic acids their effect appears to be polyvalent¹⁷. An exception among 5-azapyrimidines is represented by 5-azaorotic acid^{18,19} (oxonate), the inhibitory effect of which is increased by the combination with N-formylbiuret (Fig. 4). A similar behaviour was found with 5-fluoroorotic acid even if the two compounds differ in the mechanism of their inhibitory effect^{20,21}. A common feature of the two compounds undergoing the same metabolic transformation is an interference with the pyrimidine pathway^{18–20}. The last group of compounds, the antibacterial efficiency of which is distinctly increased by combination with N-formylbiuret are the 5-fluoroanalogues of uracil^{22,23} (Fig. 5). The antibacterial effect of other 5-halogenosubstituted analogues of uracil was not potentiated by combination with N-formylbiuret. Also the antibacterial

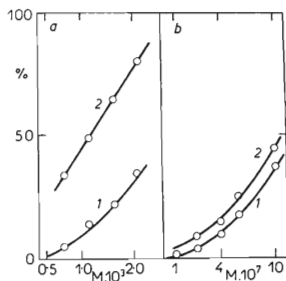


FIG. 3

Effect of N-Formylbiuret on the Antibacterial Activity of 6-Azacytidine (a) and 5-Azacytidine (b)

For details see Fig. 1.

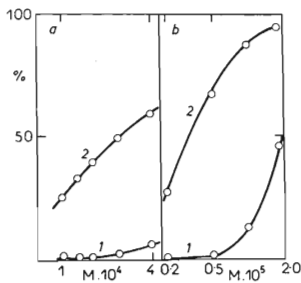


FIG. 4

Increase of Antibacterial Activity of 5-Substituted Analogues of Orotic Acid on Combination with N-Formylbiuret

a 5-Azaorotate, b 5-fluoroorotate. For details see Fig. 1.

effect of tested antifolates, uracil and cytosine arabinosides and of a number of purine analogues of nucleic acid components (6-mercaptopurine, 8-azaguanine, 8-azaadenine and adenine arabinoside) was not affected by simultaneous application with N-formylbiuret.

When evaluating the antibacterial activity of N-formylbiuret and its ability to potentiate the effects of other compounds one must take into consideration its instability. Loss of the formyl group results in the formation of buiret^{1,2} while cyclization leads to 5-azauracil² which, however, cannot potentiate the biological activity of the compounds studied. On the other hand, 5-azauracil², 5-azacytidine¹⁵ but also 5-azaorotic acid²⁴ may give rise under suitable conditions to N-formylbiuret. This fact complicates the problem why in the presence of N-formylbiuret the antibacterial effect of 6-azapyrimidines but not that of 5-azapyrimidines with the exception of 5-azaorotic acid is potentiated.

The biological activity of N-formylbiuret may be reversed by the simultaneous application of natural pyrimidine components of nucleic acids². It would thus appear that the inhibitory action of N-formylbiuret is localized in the pyrimidine pathway. However, the investigation of enzymes of the pyrimidine pathway in the presence of N-formylbiuret (Table I) did not provide evidence for this inhibition. Orotate phosphoribosyltransferase and orotidylic acid decarboxylase were not effected while carbamyl aspartotranskinase and dihydroorotase were affected to a minor degree (12.4 and 21.7% inhibition, respectively) at a concentration of 10^{-3} M N-formylbiuret. The finding may be extended by the observation that also a combination of N-formylbiuret with 6-azauridine when the biological effect is potentiated does not substantially affect the activity of enzymes of the pyrimidine pathway *in vitro*. On the basis of these findings we assume that the previously observed inhibition of *E. coli* dihydroorotase¹ cannot be considered as the only site of inhibitory action of N-formylbiuret. The fact that the biological activity of N-formylbiuret may be removed by adding pyrimidine precursors without any substantial effect on their synthesis and utilization, indicates the possibility of N-formylbiuret interfering in other areas of cell metabolism.

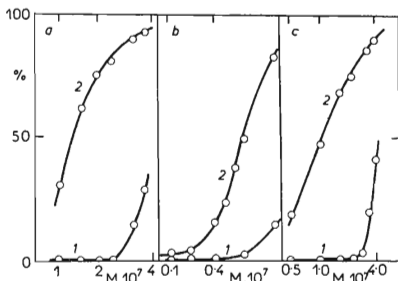


FIG. 5
Potentiation of Antibacterial Activity of 5-Fluoropyrimidines in Combination with N-Formylbiuret
a 5-Fluorouracil, b 5-fluorouridine, c 5-FUdR. For details see Fig. 1.

TABLE I

Effect of N-Formylbiuret on the Enzymes of *de novo* Pyrimidine SynthesisCalculated on the basis of 3 independent experiments using a cell-free *E. coli* extract as a source of enzyme activity.

Enzyme	Activity, $\mu\text{mol/ml}$		Inhibition %
	Control	10^{-3}M N-formylbiuret	
Carbamyl aspartotranskinase	0.73 \pm 0.12	0.64 \pm 0.08	12.4
Dihydroorotase	0.23 \pm 0.03	0.18 \pm 0.03	21.7
Orotate phosphoribosyltransferase ^a	0.049 \pm 0.008	0.051 \pm 0.006	0
Orotidylic acid decarboxylase ^a	0.038 \pm 0.005	0.038 \pm 0.007	0

^aActivity of enzymes was assayed as described earlier^{16,19}.

REFERENCES

1. Čihák A., Škoda J., Šorm F.: *Biochim. Biophys. Acta* 72, 125 (1963).
2. Čihák A., Škoda J., Šorm F.: *This Journal* 28, 3297 (1963).
3. Fučík V., Čihák A.: *Biol. Plantarum* 6, 117 (1964).
4. Link F., Rada B., Blaškovič D.: *Ann. N. Y. Acad. Sci.* 130, 31 (1965).
5. Škoda J., Čihák A., Šorm F.: *This Journal* 29, 2389 (1964).
6. Čihák A., Škoda J., Šorm F.: *This Journal* 29, 1322 (1964).
7. Holý A.: *This Journal* 31, 2973 (1966).
8. Gut J.: *Advances in Heterocyclic Chemistry* (A. R. Katritzky, Ed.), p. 189. Academic Press, New York 1963.
9. Lowry O. H., Rosebrough M. I., Farr A. L., Randal R. L.: *J. Biol. Chem.* 193, 265 (1951).
10. Reichard P., Hanshoff G.: *Acta Chem. Scand.* 10, 548 (1956).
11. Bresnick E.: *Biochim. Biophys. Acta* 26, 598 (1962).
12. Lieberman I., Kornberg A.: *Biol. Chem.* 207, 911 (1954).
13. Prusoff W. H.: *J. Biol. Chem.* 226, 901 (1957).
14. Škoda J.: *Progr. Nucl. Acid Res.* 2, 197 (1963).
15. Čihák A., Šorm F.: *This Journal* 30, 2091 (1965).
16. Veselý J., Čihák A., Šorm F.: *Biochem. Pharmacol.* 17, 519 (1968).
17. Roy-Burman P.: *Analogues of Nucleic Acid Components* (P. Rentschick, Ed.), p. 45. Springer, Berlin 1970.
18. Handschumacher R. E.: *Cancer Res.* 23, 634 (1963).
19. Čihák A., Šorm F.: *Biochim. Biophys. Acta* 149, 314 (1967).
20. Dahl J. L., Way J. L., Parks R. E., jr.: *J. Biol. Chem.* 234, 2998 (1959).
21. Wilkinson D., Čihák A., Pitot H. C.: *J. Biol. Chem.* 246, 6418 (1971).
22. Heidelberger C.: *Progr. Nucl. Acid Res.* 4, 1 (1965).
23. Mandel H. G.: *Progr. Molec. and Submolec. Biol.* 1, 82 (1969).
24. Granat P., Creasey W. A., Calabresi P., Handschumacher R. E.: *Clin. Pharmacol. Therapeut.* 6, 436 (1965).

Translated by A. Kotyk.